Yeast Influence on Volatile Composition of Wines

Andrea Antonelli,*,† Lorena Castellari,‡ Carlo Zambonelli,§ and Alberta Carnacini†

Istituto di Industrie Agrarie, University of Bologna, via S. Giacomo 7, 40126 Bologna, Italy, CATEV Srl (Centro di Assistenza Tecnologica in Enologia e Viticoltura), via Tebano 45, 48018 Faenza (RA), Italy, and Dipartimento di Protezione e Valorizzazione Agroalimentare (DIPROVAL), University of Bologna, via F.lli Rosselli, Villa Levi Coviolo, 42100 Reggio Emilia, Italy

Nine *Saccharomyces cerevisiae* and four *Saccharomyces bayanus* strains used in the production of Trebbiano wine were examined. The aim of this study is to evaluate the different abilities of various yeast strains in wine-making. The possibility of yeast discrimination on the basis of their volatile production is another possibility. Wine chemical analyses showed statistically significant differences depending on the yeast strain used. Some compounds such as 2-phenylethanol, 2-phenethyl acetate, ethyl lactate, 3-ethoxypropanol, and, to a lesser extent, diethyl succinate and propionic acid characterized examined *Saccharomyces bayanus* yeasts. Moreover, these strains did not produce any undesirable compounds, such as acetic acid and sulfur anhydride. For these reasons and because they synthesized malic acid, they could be more suitable for white wine production. The other yeasts showed great differences, which are difficult to correlate with the strain. However, some strains had peculiar characteristics, such as an uncommonly high concentration of *n*-propanol and 3-ethoxypropanol.

Keywords: Aroma compounds; 2-phenylethanol; ethyl lactate; wine volatiles; yeast strains; wine composition; gas chromatography–mass spectrometry

INTRODUCTION

Yeast influence on wine composition and quality is well-known. Besides ethanol and CO₂, the metabolism of yeasts yields a great number of byproducts, for example, glycerol, acetic acid, succinic acid, and lactic acid. Moreover, the aromatic wine properties can be deeply affected by production of higher alcohols and other volatile substances (Ingraham and Guymon, 1960; Rankine, 1967; Usseglio Tomasset, 1967; Soufleros and Bertrand, 1979; Giudici et al., 1990, Delteil and Jarry, 1991; Giudici and Zambonelli, 1992; Mateo et al., 1992; Giudici et al., 1993a,b; Castellari et al., 1994; Bertolini et al., 1996; Lema et al., 1996).

Literature on this topic shows that yeast species and, within each species, different strains have great differences in volatile compound production (Di Stefano et al., 1981; Soles et al., 1982; Houtman and Du Plessis, 1985; Cavazza et al., 1989; Herraiz et al., 1990; Bertolini et al., 1996; Gil et al., 1996; Riponi et al., 1997). Hence, the yeast-induced fermentative aroma is responsible for great differences in composition as well as in taste.

On the other hand, the variety displayed by different yeasts from a qualitative and quantitative standpoint could be used for their selection and for their taxonomic classification. In fact, *Saccharomyces cerevisiae* strains are not easily recognized from one another on the basis of their phenotypic characteristics. Therefore, killer activity, SO₂ or H₂S production, ethanol tolerance, fatty acid production, and protein composition have been studied to classify yeasts (Ribes et al., 1988; Martins et al., 1990; Rozes et al., 1992; Van Vuuren and Jacobs, 1992; Vezinhet et al., 1992). In these terms, volatile composition could help the difficult work of taxonomists (Strydom, 1985; Cavazza et al., 1989; Mateo et al., 1992).

In this paper, 13 different yeasts, which fermented the same must, were studied to focus upon significant differences between alcoholic fermentation secondary products, particularly regarding volatile composition, to evaluate yeast ability in wine-making. A contribution to the identification of possible markers of yeast species or strains is another possibility of this study.

MATERIALS AND METHODS

Organisms. Thirteen strains of *Saccharomyces* spp. from the CATEV and DIPROVAL collections, which belong to three groups with well-defined general characteristics, were used (Table 1). These characteristics are referred to a standard fermentation of industrial ripe grapes with a reducing sugar content of ~200 g/L and a total acidity content of 5–6 g/L.

(1) S. cerevisiae (SC) sensu (Vaughan-Martini and Martini, 1993). These strains are identified by the numbers 404, 1042, 5298, 6167, 6527, 7833, and 7833 2C (spore culture from 7833). These yeasts give maximum ethanol yield and minimum total level of minor compounds (i.e., higher alcohols, glycerol, acetic acid, etc.).

(2) S. cerevisiae (SC) Non- H_2S -Producing (Strains 6392 and 6842). These strains are distinctive for their high *n*-propanol and sulfite production and low-level production of other minor compounds (Giudici et al., 1990, 1993b). They have a strong stabilizing action on wines. These two groups always yield fertile spores with the exception of strain 404.

(3) S. bayanus (SB) sensu (Vaughan-Martini and Martini, 1993) (Strains 7877 3A, 11241, 11719, and 12233). These strains differ notably from S. cerevisiae strains due to their different production ratio of minor compounds: higher levels of glycerol, succinic acid, and higher alcohols (particularly

^{*} Author to whom correspondence should be addressed (fax +39 51 25 9911; e-mail antonell@agrsci.unibo.it).

[†] Istituto di Industrie Agrarie.

[‡] CATEV Srl.

[§] Dipartimento di Protezione e Valorizzazione Agroalimentare.

Table 1. Main Characteristics of Yeast Strains^a

| yeast strain | species | H ₂ S producer | acetic acid, g/L | SO ₂ , mg/L | glycerol, g/L | succinic acid, g/L | malic acid | <i>n</i> -propanol, mg/L | isobutanol, mg/L | amyl alcohols, mg/L | 2-phenyl- ethanol, mg/L |
|-----------------|---------------|------------------------------|------------------------|---------------------------|------------------|--------------------------|--------------------|-----------------------------|---------------------|---------------------------|-------------------------------|
| 404 | S. cerevisiae | yes | 0.2-0.5 | <10 | 5-8 | 0.5-0.8 | degr ^b | 20-50 | >50 | >250 | <50 |
| 1042 | S. cerevisiae | yes | 0.2 - 0.5 | <10 | <5 | < 0.5 | degr | 20 - 50 | >50 | >250 | <50 |
| 5298 | S. cerevisiae | yes | 0.2 - 0.5 | <10 | 5-8 | 0.5 - 0.8 | degr | 20 - 50 | >50 | >250 | <50 |
| 6167 | S. cerevisiae | yes | 0.2 - 0.5 | 20 - 30 | 5 - 8 | 0.5 - 0.8 | degr | 20 - 50 | 30 - 50 | 150 - 250 | <50 |
| 6527 | S. cerevisiae | yes | 0.2 - 0.5 | <10 | 5 - 8 | 0.5 - 0.8 | degr | 20 - 50 | 30 - 50 | 150 - 250 | <50 |
| 7833 | S. cerevisiae | yes | 0.2 - 0.5 | <10 | 5-8 | 0.5 - 0.8 | degr | 20 - 50 | >50 | >250 | <50 |
| 7833 2C | S. cerevisiae | yes | >0.5 | <10 | 5-8 | 0.5 - 0.8 | degr | 20 - 50 | >50 | >250 | <50 |
| 6392 | S. cerevisiae | no | 0.2 - 0.5 | 20 - 30 | 5-8 | 0.5 - 0.8 | degr | >100 | 30 - 50 | 150 - 250 | <50 |
| 6842 | S. cerevisiae | no | 0.2 - 0.5 | 20 - 30 | 5 - 8 | 0.5 - 0.8 | degr | 20 - 50 | 30 - 50 | 150 - 250 | <50 |
| 7877 3A | S. bayanus | no | < 0.2 | <10 | 5 - 8 | 0.5 - 0.8 | synth ^c | >100 | 30 - 50 | 150 - 250 | >100 |
| 11241 | S. bayanus | yes | < 0.2 | <8 | >8 | >0.8 | synth | 20 - 50 | 40 - 60 | 150 - 250 | >100 |
| 11719 | S. bayanus | yes | < 0.1 | <8 | >8 | >0.8 | synth | 20 - 50 | 40 - 60 | 150 - 250 | >100 |
| 12233 | S. bayanus | yes | < 0.1 | <8 | >8 | >0.8 | synth | 20 - 50 | 40 - 60 | 150 - 250 | >100 |

^{*a*} Data are means of various fermentations on musts and syntethic media. ^{*b*} Degradate. ^{*c*} Synthesize.

2-phenylethanol) and lower levels of acetic acid. They also synthesize malic acid instead of degrading it (Castellari et al., 1992, 1994; Bertolini et al., 1996). Strain 77877 3A is a non- H_2S -producing yeast.

Must and Fermentation. About 300 kg of cv. Trebbiano grapes was harvested at industrial maturation. Pressing with a stemmer–crusher and continuous press was carried out at the CATEV Experimental Center. The must was cold clarified (0 °C) for 48 h without any further addition, and then the clear must was heated to 18 °C and divided into 26 flasks of 3 L each.

The inoculation was carried out with 300 mL of a 3-day old statistically produced culture of the 13 yeast strain (10^5 cells/ mL), previously prepared in sterilized must, thus obtaining two flasks for each yeast. Fermentation took place at 20 °C in a controlled environment. The progress of fermentation was followed by CO₂ production. At the end of fermentation the natural sedimentation of the yeast cells (which occurred on average in 5 days) was allowed to take place, and then the settled wines were divided into subsamples for analysis.

Analysis. Ethanol was determined by densitometry at 20 °C after distillation; sulfur dioxide was determinated according to the Ripper–Schmidt method and reducing sugars were determined according to the Fehling method. Total extract of wine was measured by evaporation at 100 °C, and pH was determined by using the potentiometric method. Volatile acidity was determined by steam distillation and total acidity by titration with 0.1 N sodium hydroxide solution (phenol-phthalein as indicator) and expressed as grams per liter of tartaric acid; color absorbance was determined by a spectro-photometric method. Details of the procedure used are reported in Ough and Amerine (1987) and by EC Gazette (EU Official Gazette, 1990).

Glycerol and succinic, malic, L-lactic, and D-lactic acids were determined enzymatically with specific kits (Boehringer-Mannheim GmbH, Germany) following the procedure specified by the firm.

The higher alcohols were quantified with a Hewlett-Packard 5710 A gas chromatograph using the official AOAC method (AOAC, 1984).

GC and GC/MS volatile analyses were performed according to a previously reported method (Gerbi et al., 1992; Galletti et al., 1996).

Statistical Analysis. One-way analysis of variance (ANO-VA) and Tukey's honest significant differences were performed by a Statgraphics package (STSC, Rockville, MD) on the wine volatiles only.

RESULTS AND DISCUSSION

The must, the composition of which is reported in Table 2, was rapidly fermented by all strains used, and the process ended within 20 days. In the same table, wine composition is also displayed. During the process, yeasts showed almost no differences, as already reported for many of them (Riponi et al., 1997). As expected, SB strains produced lower ethanol content as a consequence of higher secondary products accumulation (i.e., malic acid and volatiles).

Wine composition showed some remarkable differences in fixed compounds. As an example of the different yeast metabolisms, glycerol varied from 3.58 g/L (strain 6392) to 6.95 mg/L (strain 78773A).

Despite the small differences among ethanol concentrations, some statistical differences were present. On the other hand, acetic acid showed a wider range of variation. All SB strains produced low quantities of this undesirable substance. In contrast, some SC strains (6527 and 78332C) reached the highest quantities (Table 2).

No sample underwent malolactic fermentation, because malic acid was present in comparable quantities in must as well as in wines and the L-lactic acid amount was below 0.01 g/L, whereas SB synthesized malic acid, thus reaching 2.09 mg/L (7877 3A). Less remarkably, succinic acid showed the same behavior, even if strain 6527, a SC yeast, displayed amounts comparable to those of SB strains.

All volatile substances (Tables 2 and 3) displayed statistical differences among each strain, except for methanol, which is not a fermentation product, and benzyl alcohol.

Acetaldehyde showed wide fluctuations regardless of yeast group, but its concentration is affected by SO₂ content. In fact, linear regression analysis showed a good agreement between these two parameters (y = 29 + 1.16x, r = 0.73, p > 0.99).

Ethyl acetate is strongly influenced by acetic acid content. As a consequence, low acetic acid producers had the lowest ester content. Anyway, this relationship is not so strict, because of the importance of the different yeast esterase activities (Dell'Oro and Delfini, 1992). In any case, yeasts formed more ethyl acetate than would be expected from the law of mass action (Rapp, 1988). In regression analysis of ethyl acetate versus acetic acid (y = 14 + 27x; r = 0.53; p > 0.99), the large data dispersion confirms this fact. On the contrary, acetaldehyde versus SO₂ is simply a chemical reaction, and hence it is mainly influenced by reagent concentrations.

Higher alcohols (*n*-propanol, isobutanol, amyl alcohols) were in good agreement with what has been previously published both on the same strains (Giudici et al., 1990; Castellari et al., 1994; Bertolini et al., 1996) and, more in general with literature data (Giudici et al., 1985, 1993a; Kunkee and Vilas, 1994). However, two

| ion ^a |
|------------------|
| positi |
| Com |
| Wine |
| સં |
| le |
| ab |
| |

| | | must | 404 | 1042 | 5298 | 6167 | 6527 | 7833 | 7833 2C | 6392 | 6842 | 7877 3A | 11241 | 11719 | 12233 |
|--|----------------------|------------------------|-------------------------------------|--|---------------------------|--|---------------------------|-----------------|----------------|--------------|--------------|----------------|--------------|-------------------|--------------|
| reducing sugar pH | g/L | 177 3.14 | 1.10 3.22 | 1.06 3.22 | $^{<1.00}_{3.24}$ | 2.24 3.19 | 1.55 3.18 | 1.03 3.14 | 2.01 3.19 | 1.25 3.18 | 1.64 3.15 | 1.65 3.06 | 3.30 3.14 | $^{<1.00}_{3.09}$ | 2.30 3.08 |
| total acidity ^b | g/L | 3.86 | 5.65 0.21 | 5.70 | 5.53 | 5.10 0.29 | 6.10 0.46 | 6.05 | 5.43 | 5.10 | 5.45 | 7.48 | 6.35 | 7.78 | 7.90 |
| D-lactic acid | رت مرتاح | pu | 0.147 | 0.188 | 0.077 | 0.105 | 0.107 | 0.16 | 0.095 | 0.120 | 0.166 | 0.277 | 0.275 | 0.334 | 0.262 |
| malic acid | g/L | 1.20 | 0.93 | 1.07 | 0.77 | 0.94 | 0.84 | 0.97 | 0.75 | 0.99 | 0.97 | 1.88 | 1.43 | 1.76 | 2.00 |
| succinic acid | g/L | pu | 0.76 | 0.70 | 0.69 | 0.54 | 1.17 | 1.07 | 0.70 | 0.61 | 0.78 | 1.63 | 1.26 | 1.96 | 1.85 |
| glycerol | g/L | pu | 5.19 | 5.41 | 5.22 | 4.12 | 4.90 | 5.82 | 6.52 | 3.97 | 4.63 | 6.85 | 5.71 | 6.59 | 6.71 |
| total SO_2 | mg/L | 60 | 46 | 21 | 23 | 58 | 42 | 29 | 24 | 57 | 56 | 50 | 19 | 26 | 29 |
| acetic acid | g/Ľ | pu | 0.18C DEF | 0.18 E | 0.33F | 0.21 CDE | 0.38F | 0.22E | 0.58 G | 0.15 CDE | 0.14 BCD | 0.12 ABC | 0.09 AB | 0.12 ABC | 0.07A |
| ethanol | vol % | pu | 10.71 CD | 10.87 E | 10.85 DE | 10.72 CDE | 10.63 BC | 10.70 BC | 10.72 CDE | 10.85 DE | 10.83 DE | 10.67 CD | 10.53 AB | 10.37 A | 10.38 A |
| acetaldehyde | mg/L | pu | 82 CDE | 49 AB | 37 A | 102 E | 76 CD | 59 ABC | 95 DE | 95 DE | 82 CDE | 104 E | 68 BC | 44 A | 60 ABC |
| ethyl acetate | mg/L | pu | 21 CD | 22 CD | 20 CD | 17 BC | 26 DEF | $31 \mathrm{F}$ | 26 DEF | 29 EF | 23 CDE | 16 BC | 11 AB | 11 AB | 9 A |
| methanol | mg/L | pu | 54 ns | 57 ns | 51 ns | 61 ns | 52 ns | 57 ns | 58 ns | 57 ns | 56 ns | 59 ns | 59 ns | 55 ns | 59 ns |
| n-propanol | mg/L | pu | 31 BC | 18 A | 17 A | 35 CD | 17 A | 19 A | 11 A | 182 E | 45 D | 20 AB | 16 A | 13 A | 13 A |
| isobutanol | mg/L | pu | 79 BCD | 104 DEF | $123 \mathrm{F}$ | 43A | 92 CDE | 132 F | 111 EF | 45 A | 56 AB | 73 ABC | 47 A | 72 ABC | 45 A |
| amyl alcohols | mg/L | pu | 215 BC | 248 BC | 231 BC | 160 ABC | 189 ABC | 253 BC | 212 BC | 150 ABC | 169 ABC | 144 ABC | 67 A | 130 AB | 137 AB |
| ^a For ethano (Tukey's test); : | l, acetic ns, not | s acid, s significa | und other major ant; nd, not det | r volatiles, termined. ¹ | data are m As tartaric | two ruaries of two ruaries acid. ^c As ace | eplications. tic acid. | ANOVA rea | sults are also | reported. Nu | mbers with d | ifferent capit | al letters d | liffer at p < 0 | .01 level |

| Table 3. Volatile C | ompositio | n (Microgr | ams per Lit | er) Produce | d by the Dif | ferent Yeas | ts Strains ^a | _ | | | | | |
|---------------------------------|-----------------------|-----------------------|--------------------|----------------|--------------|-----------------------|-------------------------|-------------------|--------------------|------------------|-----------------------|-----------|-----------------------|
| | 404 | 1042 | 5298 | 6167 | 6527 | 7833 | 7833 2C | 6392 | 6842 | 7877 3A | 11241 | 11719 | 12233 |
| isobutyl acetate | 242 B | tr A | 585 C | 173 AB | 543 C | 334 B | 230 B | 233 B | 248 B | 255 B | 2399 D | 275 B | 207 B |
| 2- and 3-methylbutyl acetate | 873 D | $1627 \mathrm{F}$ | 971 DE | 424 B | 631 BC | 1153 E | 659 C | 606 BC | 1015 DE | 463 BC | 139 A | 170 A | 172 A |
| n-butanol | 1069 B | 1666 CD | 1883 D | 2856 E | 868 B | 1490 C | 1044 B | 1423 C | 1376 C | 421 A | 923 B | 346 A | $440\mathrm{A}$ |
| amyl acetate | 11 AB | 19 BC | $10\mathrm{AB}$ | 32 C | 20 BC | 11 AB | tr A | tr A | $13 \mathrm{AB}$ | 12 AB | tr A | 8 AB | tr A |
| ethyl hexanoate | 367 E | $541 \mathrm{F}$ | 341 DE | 214 ABC | 223 ABC | 375 E | 167 A | 365 E | 281 CD | 247 BC | $201 \mathrm{AB}$ | 250 BC | 174 A |
| n-amylic alcohol | 69 DEF | 41 A | 67 CDEF | 65 BCDE | 51 ABC | $82 \mathrm{F}$ | $50 \mathrm{AB}$ | 51 ABC | 55 ABCD | 56 ABCDE | 71 EF | 65 BCDE | $44 \mathrm{A}$ |
| hexyl acetate | 53 CD | 70 E | 39 BC | 24 B | 27 B | 57 DE | 30 B | 37 B | $60 \mathrm{DE}$ | 61 DE | $\operatorname{tr} A$ | tr A | 23 B |
| 3-methyl-1-pentanol | 72 DE | 68 CDE | 55 ABCD | 39 A | 47 ABC | 95 F | $40\mathrm{A}$ | 62 BCDE | 79 EF | 54 ABC | 42 AB | 43 AB | $40 \mathrm{A}$ |
| ethyl lactate | 19699 AB | 17510 AB | 12365 AB | 16865 AB | 14271 AB | 19442 AB | 10592 A | 14561 AB | 21726 B | 36602 C | 34190 C | 36253 C | 33857 C |
| 1-hexanol | 1417 B | 1131 AB | 1259 B | 1135 AB | 960 A | 1281 B | 1138 AB | 1227 AB | 1291 B | 1235 AB | 1314 B | 1180 AB | 1171 AB |
| trans-3-hexen-1-ol | 49 B | 35 A | $41 \mathrm{AB}$ | 40 AB | 34A | $41\mathrm{AB}$ | 37 AB | 42 AB | 50 B | 38 AB | $43 \mathrm{AB}$ | 41 AB | 37 AB |
| 3-ethoxy-1-propanol | 2117 C | tr A | $303\mathrm{AB}$ | 1994 C | 246 AB | $174 \mathrm{AB}$ | 259 AB | 22553 E | 4574 D | 809 B | 353 AB | 391 AB | $655 \mathrm{AB}$ |
| cis-3-hexen-1-ol | 78 E | 62 ABC | 73 CDE | 62 ABC | 59 AB | 73 CDE | $56 \mathrm{A}$ | 64 ABCD | 77 DE | 71 BCDE | 73 BCDE | 63 ABCD | 64 ABCD |
| ethyl octanoate | 626 BC | 1041 D | 817 C | 352 A | 456 AB | $499 \mathrm{AB}$ | 630 BC | 620 BC | 434 AB | 472 AB | 344 A | 369 A | 367 A |
| ethyl 3-OH-butanoate | $\operatorname{tr} A$ | $\operatorname{tr} A$ | 327 C | tr A | 241 B | $\operatorname{tr} A$ | $\operatorname{tr} A$ | 316 C | 828 D | 205 B | $\operatorname{tr} A$ | tr A | $\operatorname{tr} A$ |
| propionic acid | 66 ABC | 49 A | 171 H | 82 CDEF | 90 EF | 74 BCDE | 94 F | 65 AB | 67 BCD | 132 G | 131 G | 159 H | 127G |
| isobutanoic acid | 8671 ABC | 7794 A | $20849 \mathrm{F}$ | 10889 ABCD | 11814 CD | 9009 ABC | 12238 D | 10437 ABCD | 8337 AB | 12489 D | 16440 E | 11321 BCD | 12685 D |
| γ -butyrolactone | 33310 A | 30327 A | 60484 CDE | 44911 B | 61060 DE | 54579 CD | 36357 A | 34164 A | 34154 A | 61985 DE | 67773 E | 52745 BC | 58313 CD |
| ethyl decanoate | 1545 AB | 1560 AB | 2217 BC | 1189 A | 1372 A | 1756 ABC | $1270 \mathrm{A}$ | $1296 \mathrm{A}$ | 1087 A | 1186 A | 3481 D | 2357 C | 1479 A |
| diethyl succinate | 328 CDE | 222 ABC | 269 BCD | 134 A | 349 CDE | 320 CDE | $145 \mathrm{AB}$ | 185 AB | $154 \mathrm{AB}$ | 443 EF | $414 	ext{ E}$ | 566 F | 568 F |
| 3-methylthiopropanol | 11592 E | 11620 E | 9847 E | 1289 AB | 5806 D | 10059 E | 5631 D | $962 \mathrm{A}$ | 3256 BC | 2682 ABC | 4582 CD | 4628 CD | 4246 CD |
| ethyl 4-OH-butanoate | 520 A | 2336 DEF | 2785 FGH | 2727 FG | 2625 EFG | 2474 DEFG | 1325 B | 1716 BC | 2091 CD | 2200 CDE | 3282 H | 2512 DEFG | 2940 GH |
| phenethyl acetate | 352 ABC | 526 CD | 420 BCD | 166 A | 329 ABC | 317 AB | 324 ABC | 345 ABC | 376 BC | 969 F | 586 DE | 1453 G | 756 E |
| hexanioc acid | 3612 FG | 4242 G | 3608 FG | 2716 BCD | 2969 CDEF | 2798 CDE | $1545 \mathrm{A}$ | 3028 CDEF | 2481 BC | 2501 BC | 3408 EF | 3381 DEF | 2086 AB |
| benzyl alcohol | 466 ns | 309 ns | 359 ns | 400 ns | 349 ns | 296 ns | 317 ns | 592 ns | $517 \mathrm{ns}$ | 283 ns | 373 ns | 333 ns | 209 ns |
| phenethyl alcohol | 92327 CDE | 60012 AB | 82071 BCD | 54096 A | 56088 AB | 54622 A | 70444 ABC | 63126 AB | 63837 AB | 108051 DE | 94587 CDE | 107332 DE | 111173 E |
| octanoic acid | 5417 F | 7775 G | 4528 DEF | 4539 DEF | 3422 BC | 4064 CDE | $1836 \mathrm{A}$ | 4464 DEF | 5197 F | 4788 EF | 3843 BCDE | 3704 BCD | 3078 B |
| decanoic acid | 1924 F | 1809 EF | 1067 BCD | 1596 DEF | 995 BC | $1326 \mathrm{F}$ | 387 A | 1153 BCD | 2921 G | 2862 G | 2498 G | 1485 CDEF | 1506 CDEF |
| ^a Data are means e | of two replic | ations. ANG | OVA results ; | are also repor | ted. Number | s with differe | ent capital l | etters differ | at p < 0.01 le | vel (Tukey's tes | t); ns, not sign | nificant. | |



Figure 1. Total aroma amount for each yeast (higher alcohols are not included).

strains (12233 and 11241) were generally poor in higher alcohols content, and yeast 7833 was the richest. *n*-Propanol was always the less abundant alcohol, except for strain 6392. In this case it was the main alcohol. Finally, unexpected low quantities of propanol (7877 3A) and amyl alcohols (11241) are less frequently reported. Herraiz et al. (1990) reported 37.24 mg/L for the sum of amyl alcohols, whereas Gil et al. (1996) reported even lower contents for these substances with apiculate yeasts. The presence of significantly different quantities of 1-butanol was also documented by the latter author. Our samples show a small but sharp difference of this analyte between SC and SB.

The sum of Table 3 substances (Figure 1) indicated SB strains as the greatest aroma compound producers. Ethyl lactate and 2-phenylethanol justify these differences and characterized SB strains. These yeasts, in fact, are known as 2-phenylethyl producers, even if our data were particularly high compared with those of the literature (Gil et al., 1996). Also, Di Stefano et al. (1981) reported a concentration 10-fold higher of this alcohol in Saccharomyces uvarum, now belonging to the SB group (Vaughan-Martini and Martini, 1993) versus SC. Moreover, in the same strains this higher 2-phenylethanol content yield higher phenethyl acetate amount, which was not significantly correlated with acetic acid concentration. In fact, no significant correlation between these two substances was found (r = -0.17; F = 1.69; ns).

The higher ethyl lactate concentrations of SB can be related to the malic acid production typical of these yeasts. Because no malolactic fermentation had occurred, all ethyl lactate was due to yeast activity. In addition, the higher quantities of lactic and succinic acid developed by SB strains were responsible for their ethyl ester accumulation. Several authors have reported lower concentrations of this hydroxyl ester (Di Stefano et al., 1981; Herraiz et al., 1990; Gil et al., 1996), but Dubois (1994) in his review reported quantities up to 534 mg/L for ethyl lactate.

It is worth mentioning the very high quantity of 3-ethoxypropanol typical of yeast 6392. The following strain in order of abundance (6842) reached only onefifth of the previous one. As already discussed, these two strains showed a particularly high propanol content, the relation of which with 3-ethoxypropanol is evident. Di Stefano et al. (1981) reported this compound as typical of certain yeasts, and recently some of us detected a high increase of this alcohol during flor aging of Vernaccia di Oristano wine. Other authors (Herraiz et al., 1990) found comparable amounts of this alcohol in *Torulospora delbruekii* but with no correlation with *n*-propanol content.

Some SC strains (404, 1042, 5298, 7833) produced >10 mg/L of 3-methylthio-1-propanol. The other sub-

stances were evenly distributed among yeast strains without any evident relationship. Six carbon atom alcohols, which mainly derive from enzymatic cleavage of fatty acids during juice extraction, were slightly affected by yeast. However, previous papers by some of us (Galletti et al., 1996; Carnacini et al., 1997) reported an increase of these substances due to flor yeast metabolism.

Acetates and short-chain fatty acid esters are very important for the sensory characteristics of white wines (Soles et al., 1982). The anomalous value for isobutyl acetate in strain 11241 could be interesting, but it needs further confirmation. In fact, the other acetates of this sample were of the same order of magnitude compared to the other samples, and its low isobutanol amount makes isobutyl acetate datum doubtful. Apart from these considerations, strains 11241 and 1042 were the best acetate producers, whereas strain 6167 was the poorest. Despite the influence of acetic acid and the relative alcohol on acetate yields, there was no statistical correlation among them. These observations are in good agreement with the findings of other authors (Soles et al., 1982; Houtman and Du Plessis, 1985; Cavazza et al., 1989; Gil et al., 1996). Only phenethyl acetate showed a linear regression with its related alcohol (y =297 + 0.01x, r = 0.75; p > 0.99). The absence of any correlation between acetic acid and acetates indicated that acids were not the limiting factor for acetate production. In fact, SB, which presented minimum levels of acetic acid, had the highest acetate contents. As already supposed for ethyl acetate, the esterase activity and its selectivity are the main reasons for acetate content.

The most efficient acetate producers synthesized also the highest content of short-chain fatty acid esters (ethyl hexanoate, ethyl octanoate, and ethyl decanoate). However, some strains had different behaviors to the different homologues. Strain 11241 produced mainly ethyl decanoate, as did strain 1042 ethyl hexanoate. A straight correlation is shown between total fatty esters amount and total fatty acids amount (y = 154 + 0.11x, r = 0.61; p > 0.99). Compared to small fatty acid variations in each sample, the great ethanol excess makes this procedure meaningless.

Isobutanoic acid was the main short-chain fatty acid. Its content was often greater than the sum of the remaining ones. Propionic acid was the poorest, but SB and SC 5298 accumulated it more effectively than other yeasts. Dubois (1994) reported low butanoic acid quantities and indicated octanoic acid as the main compound within this class. The others were in comparable amounts. This discrepancy could be due to the different extraction techniques.

From a quantitative point of view hydroxyl acid ethyl esters were well represented. Ethyl lactate was the most abundant, and it is well represented in every sample with some important differences (see above), but only five strains produced appreciable amounts of ethyl 3-hydroxybutanoate. On the contrary, ethyl 4-hydroxybutanoate was ubiquitous, with concentrations from 1 to ~4 mg/L, except for strain 404, which produced only 520 μ g/L. γ -Butyrolactone, 10-fold more concentrated than ethyl 4-hydroxybutanoate, displayed almost the same distribution. In fact, they come from a common precursor: glutamic acid through 4-hydroxybutanoic acid (Dufossé et al., 1994). The free acid is not present because it lactonizes spontaneously or it is esterified as



Figure 2. Cluster analysis of the 13 yeasts performed using all volatile substances.

for the other fatty acids. Lactone is the favorite form of accumulation. A linear correlation of γ -butyrolactone versus 4-hydroxybutanoate (y = 28906 + 8x, r = 0.52, p > 0.99) underscores this fact.

In cluster analysis (Figure 2), the first nine strains of the dendrogram are SC yeast, and the last four are SB. The evident presence of the two main groups is a consequence of their volatile production (Tables 2 and 3), as outlined before; it also completes what has been previously reported by some of us on the same yeasts, regarding some nonvolatile parameters and optimal temperature of fermentation (Castellari et al., 1994; Bertolini et al., 1996).

Strain 404, widely used by Italian wine-makers, seems to be completely different from all the other yeasts. This yeast produces only a few nonvital spores and, due to this fact, it is difficult to classify it from a taxonomic standpoint. Some authors reported these sterile yeasts to be natural interspecific hybrids (Naumov et al., 1994), haiving behavior apart from the parent species.

The strain 7833 2C is a culture obtained from a single spore from 7833 strain. The latter is a heterozygote genotype for higher alcohol production, and all of its single-spore cultures show a great variability of this parameter (Zambonelli, 1991). This fact justifies the distance between them in the diagram.

On the contrary, the position of strain 5298 is taxonomically more difficult to explain.

CONCLUSIONS

The vinification trials with 13 different yeasts (species as well as strains) showed their influence on the volatile composition of wine.

The outcomes clearly indicate some differences between SB and SC strains. The former produced higher volatile content, mainly phenylethanol and ethyl lactate. The latter are less characterized but exhibited some peculiarity, such as high 3-ethoxypropanol or 3-methylthiopropanol concentrations. Also, ethyl esters vary within a large range. Some of these differences are so sharp that it could suggest a distinction between producers and nonproducers. However, this last observation needs further confirmation. Finally, SB strains fit very well into the white wie-making process not only for their volatile production but also for their low acetic acid quantities and for malic acid accumulation.

On the other hand, at the moment, the great differences among yeasts do not allow sure taxonomic considerations based upon volatile composition. Further studies and the use of other analytical parameters could help to achieve this purpose.

LITERATURE CITED

- AOAC. Alcohols (Higher) and Ethyl Acetate in Distilled Liquors (9.091); AOAC: Washington, DC, 1984; p 184.
- Bertolini, L.; Zambonelli, C.; Giudici, P.; Castellari, L. Higher alcohol production by cryotolerant *Saccharomyces* strains. *Am. J. Enol. Vitic.* **1996**, *47*, 343–345.
- Carnacini, A.; Antonelli, A.; Galletti, G. C.; Natali, N.; Farris, G. A. Quantitative change of some volatile components in Vernaccia di Oristano (a Sardinian Sherry-like wine) during maturation. J. Agric. Food Chem. 1997, 45, 2225–2228.
- Castellari, L.; Pachioli, G.; Zambonelli, C.; Tini, V.; Grazia, L. Isolation and initial characterization of cryotolerant *Saccharomyces* strains. *It. J. Food Sci.* **1992**, *3*, 179–186.
- Castellari, L.; Ferruzzi, M.; Magrini, A.; Giudici, P.; Passarelli, P.; Zambonelli, C. Unbalanced wine fermentation by cryotolerant vs. non-cryotolerant Saccharomyces strains. Vitis 1994, 33, 49–52.
- Cavazza, A.; Versini, G.; Dalla Serra, A.; Romano, F. Characterization of six *Saccharomyces cerevisiae* strains on the basis of their volatile composition production, as found in wines of different aroma profiles. *Yeast* **1989**, *5* (Special Issue), 163S–167S.
- Dell'Oro, V.; Delfini, C. Esterasi nei lieviti. *Vini d'Italia* **1992**, *34* (3), 49–54.
- Delteil, D.; Jarry, J. M. Effects caractéristiques de deux souche de levures oenologiques sur la composition en élements volatils de vins de Chardonnay. *Rev. Fr. Oenol.* **1991**, *132*, 41–46.
- Di Stefano, R.; Ciolfi, G.; Delfini, C. Composti volatili prodotti dai lieviti. *Riv. Vitic. Enol.* **1981**, *34*, 342–355.
- Dubois, P. Les arômes des vins et leurs défauts. *Rev. Fr. Oenol.* 1994, 145, 27–40.
- Dufossé, L.; Latrasse, A.; Spinnler, H.-E. Importance des lactones dans les aromes alimentaires: structure, distribution, propriétés sensorielle et biosynthése. *Sci. Aliment.* **1994**, *14*, 17–50.
- EU Official Gazette. L 272, Luxembourg, Oct 3, 1990.
- Galletti, G. C.; Carnacini, A.; Antonelli, A.; Farris, G. A. Volatile composition of Vernaccia di Oristano sherry-like wine as affected by biological ageing. *J. Sci. Food Agric.* **1996**, *70*, 44–50.
- Gerbi, V.; Zeppa, G.; Carnacini, A. Rapid extraction of volatile compounds in wine and vinegar using Extrelut resin. *It. J. Food Sci.* **1992**, *4*, 259–267.
- Gil, J. V.; Mateo, J. J.; Jiménez, M.; Pastor, A.; Huerta, T. Aroma compounds in wine as influenced by apiculate yeasts. *J. Food Sci.* **1996**, *61*, 1247–1249, 1266.
- Giudici, P.; Zambonelli, C. Biometric and genetic study on acetic acid production for breeding of wine yeasts. *Am. J. Enol. Vitic.* **1992**, *43*, 370–374.
- Giudici, P.; Silvestroni, O.; Ghidoni, G. Influenza della fonte azotata sulla attività di ceppi di lievito con riferimento alla formazione di alcoli superiori. *Vignevini* **1985**, *12* (4), 51– 57.
- Giudici, P.; Romano, P.; Zambonelli, C. A biometric study of higher alcohol production from *Saccharomyces cerevisiae*. *Can. J. Microbiol.* **1990**, *36*, 61–64.
- Giudici, P.; Altieri, C.; Gambini, S. Influenza del ceppo di lievito sui prodotti minoritari della fermentazione alcolica. *Ind. Bev.* **1993a**, *22*, 303–306
- Giudici, P.; Zambonelli, C.; Kunkee, R. E. Increased production of *n*-propanol in wine by yeast strains having an impaired ability to form hydrogen sulfide. *Am. J. Enol. Vitic.* **1993b**, *44*, 17–21.
- Herraiz, T.; Reglero, G.; Herraiz, M.; Martin-Alvarez, P. J.; Cabezudo, M. D. The influence of the yeast and type of culture on the volatile composition of wines fermented without sulfur dioxide. *Am. J. Enol. Vitic.* **1990**, *41*, 313– 318.
- Houtmann, A. C.; Du Plessis, C. S. Influence du cépage et la souches de levure sur la vitesse de de fermentation et sur

la concentration des composants volatils du vin. *Bull. O. I. V.* **1985**, *648–649*, 235–246.

- Ingraham, J. L.; Guymon, J. F. The formation of higher aliphatic alcohols by mutant strains of *Saccharomyces cerevisiae*. Arch. Biochem. Biophys. **1960**, 88, 157–166.
- Kunkee, R. E.; Vilas, M. R. Toward an understanding of the relationship between yeast strain and flavor production during vinification: Flavor effect in vinification of a non distinct variety of grapes by several strains of wine yeast. *Wien Wissen. Technol.* **1994**, *49* (1), 40–45.
- Lema, C.; Garcia-Jares, C.; Oriols, I.; Angulo, L. Contribution of *Saccharomyces* and non-*Saccharomyces* populations to the production of some components of some components of Albariño wine aroma. *Am. J. Enol. Vitic.* **1996**, *47*, 206– 216.
- Martins, G.; Montrocher, G. R.; Poncet, S. Différenciation rapide de souches de levures de vinification par une étude de caractérs morphologiques et biochimiques. *Rev. Fr. Oenol.* **1990**, *129*, 35–43.
- Mateo, J. J.; Jimenez, M.; Huerta, T.; Pastor, A. Comparison of volatiles produced by four *Saccharomyces cerevisiae* strains isolated from Monastrell musts. *Am. J. Enol. Vitic.* **1992**, 43, 206–209.
- Naumov, G. I.; Nikonenko, T. A.; Kondrateva, V. I. A taxonomic identification of Saccharomyces from yeast-geneticstokes-center of University-California. *Genetika* **1994**, *30*, 45–48.
- Ough, C. S.; Amerine, M. A. In *Methods for Analisis of Must and Wines*; Wiley: New York, 1988.
- Rankine, B. C. Formation of higher alcohols by wine yeasts, and relationship to taste and thresholds. *J. Sci. Food Agric.* **1967**, *18*, 583–589.
- Rapp, A. Wine aroma from gas chromatographic analysis. In Wine Analysis; Linskens, H. F., Jackson, J. F., Eds.; Springer-Verlag: Berlin, 1988.
- Ribes, P.; Nomdedeus, L.; Laueichesse, D.; Leaute, R. Methodes statistiques à la differenciation et la reconnaissance

de souches de Saccharomyces cerevisiae. Rev. Fr. Oenol. 1988, 114, 35-42.

- Riponi, C.; Carnacini, A.; Antonelli, A.; Castellari L.; Zambonelli, C. Influence of yeast strain on the composition of wines for the production of brandy. *J. Wine Res.* **1997**, *8*, 41–55.
- Rozes, N.; Garcìa-Jares, C.; Larue, F.; Lonvaud-Fenel, A. Differentiation between fermentating and spoilage yeasts in wine by total free fatty acid analysis. *J. Sci. Food Agric.* **1992**, *59*, 351–357.
- Soles, R. M.; Ough, C. S.; Kunkee, R. E. Ester concentrations differencies in wine fermented by various species and strains of yeasts. *Am. J. Enol. Vitic.* **1982**, *34*, 94–98.
- Soufferos, E.; Bertrand, A. Role de la "souche de levure" dans la production des substances volatiles en cours de la fermentation du jus de raisin. *Connaiss. Vigne Vin* **1979**, *13*, 181–198.
- Strydom, M. Caractérisation de quatorze souches de levures par leurs capacités de fermentation et par la composition et qualité du vin produit. *Bull. O. I. V.* **1985**, *648–649*, 218–227.
- Usseglio Tomasset, L. L'alcol β -feniletilico nei vini. *Riv. Viticol. Enol.* **1967**, *20*, 10–35.
- Usseglio Tomasset, L. L'acetato d'etile e gli alcoli superiori nei vini. *Riv. Viticol. Enol.* **1971**, *6*, 236–253.
- Van Vuuren, H. J. J.; Jacobs, J. Killer yeasts in the wine industry: a review. Am. J. Enol. Vitic. **1992**, 43, 119–123.
- Vaughan-Martini, A.; Martini, A. A taxonomic key for the genus Saccharomyces. Syst. Appl. Microbiol. 1993, 16, 113– 119.
- Vezinhet, F.; Hallet, J. N.; Valade, M.; Poulard, A. Ecological survey of wine yeast strains by molecular method of identification. *Am. J. Enol. Vitic.* **1992**, *43*, 83–86.

Received for review July 8, 1998. Revised manuscript received December 15, 1998. Accepted December 18, 1998.

JF9807317